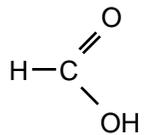
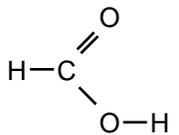
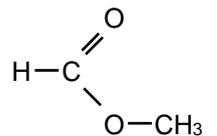
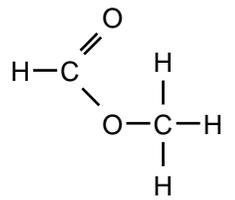
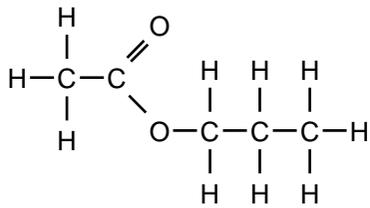
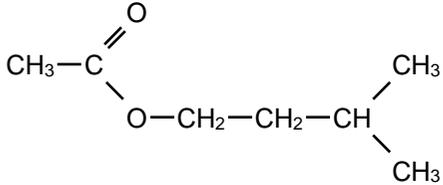
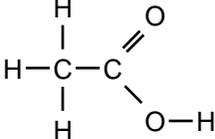
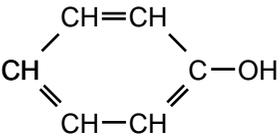
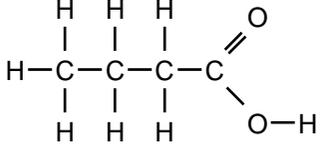
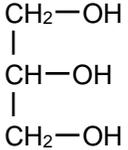


Compléter :

Formule brute	Formule semi-développée	Formule développée	Groupe fonctionnel
Acide méthanoïque : (=acide formique) $C_2H_2O_2$	H-COOH 		
Méthanoate de méthyle : (Formiate de méthyle) $C_2H_4O_2$	H-COO-CH ₃ 		
Ethanol : C_2H_6O			
Acétate de propyle : $C_5H_{10}O_2$			
Butan-2-ol : $C_4H_{10}O$	$\begin{array}{c} \text{CH}_3-\text{CH}-\text{CH}_2-\text{CH}_3 \\ \\ \text{OH} \end{array}$		

<p>Acétate de 3-méthylbutyle :</p> <p>$C_7H_{14}O_2$</p>			Ester
<p>Acide éthanoïque : (=acide acétique)</p> <p>$C_2H_4O_2$</p>			Acide carboxylique
<p>Phénol</p> <p>C_6H_6O</p>			Alcool
<p>Acide butanoïque : (Acide butyrique)</p> <p>$C_4H_8O_2$</p>			Acide carboxylique
<p>Glycerol</p> <p>$C_3H_8O_3$</p>			Alcool x 3

Acide nonanoïque : (Acide pelargonique)		$ \begin{array}{ccccccc} & \text{H} & \text{H} & \text{H} & \text{H} & \text{H} & \text{O} \\ & & & & & & // \\ \text{H} & - \text{C} \\ & & & & & & \backslash \\ & \text{H} & \text{H} & \text{H} & \text{H} & \text{H} & \text{O} - \text{H} \end{array} $	Acide carboxylique
Tributyrates de glycéryle $\text{C}_{15}\text{H}_{26}\text{O}_6$	$ \begin{array}{c} \text{O} \\ \\ \text{CH}_3 - \text{CH}_2 - \text{CH}_2 - \text{C} \\ \backslash \\ \text{O} - \text{CH} \\ \\ \text{CH}_2 - \text{O} \\ \\ \text{O} \\ \\ \text{C} - \text{CH}_2 - \text{CH}_2 - \text{CH}_3 \\ / \\ \text{O} \\ \\ \text{C} - \text{CH}_2 - \text{CH}_2 - \text{CH}_3 \end{array} $		Ester x3